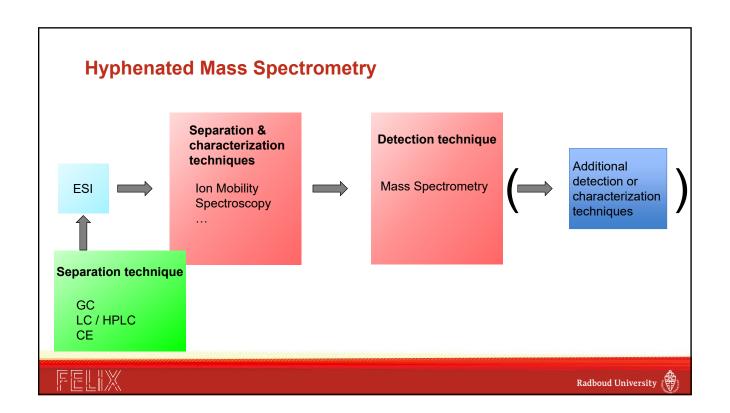
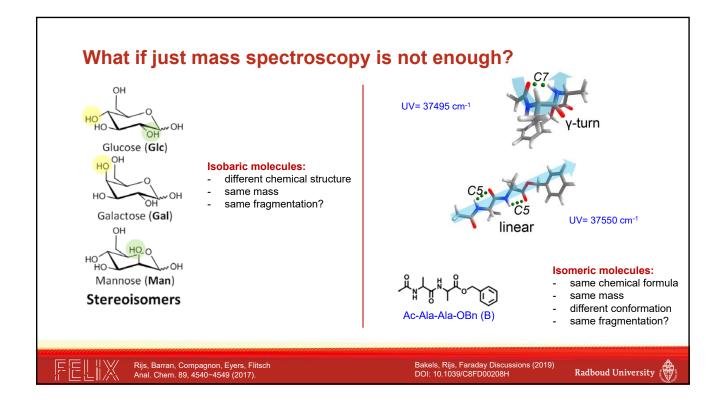
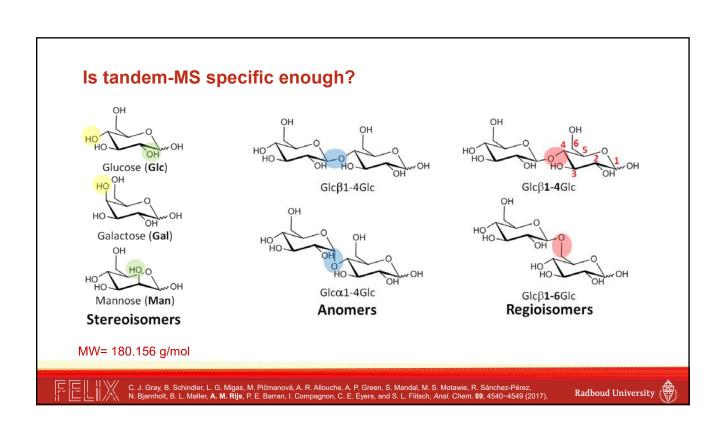
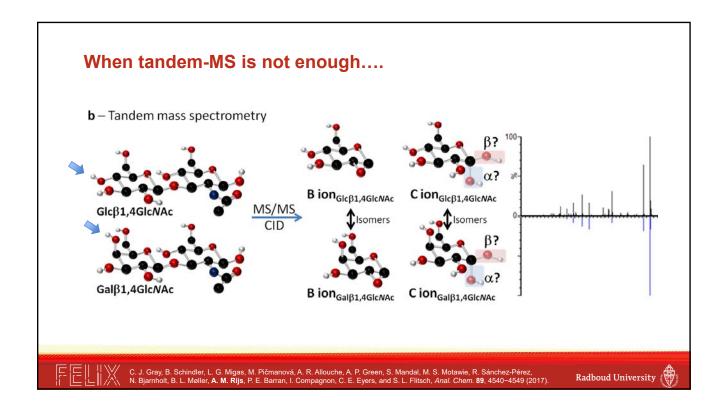
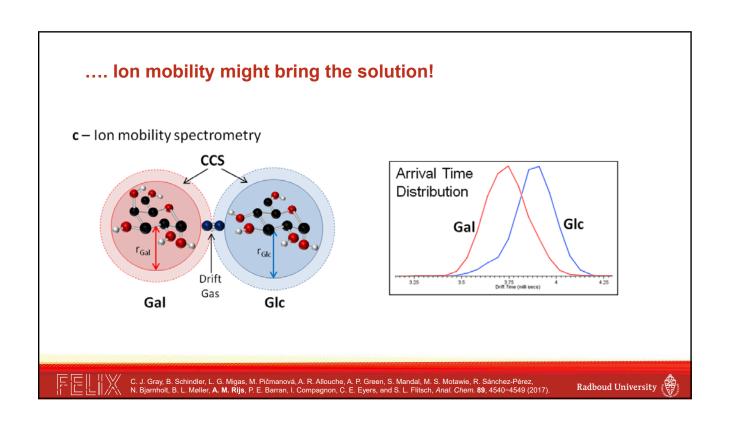
Advances in Mass Spectrometry: ION MOBILITY Background literature: • Wyttenbach and Bowers, Top Curr Chem 225, 207-232 (2003) • Clemmer et al., Annu Rev Anal Chem 1, 293-327 (2008) • Gabelica and Marklund, Curr Opinion Chem Biol 42, 51-59 (2018) FELIX Laboratory E: a.rijs@science.ru.nl W: www.ru.nl/molphys











- 1) Background into IM-MS (what?, history, fundamental principles)
- 2) Obtaining mobilities and CCS values
- 3) Instrumentation
- 4) Various types of ion mobility
- 5) Separation or characterization?
- 6) Applications / papers

Background literature:

- book
- Wyttenbach and Bowers,
 Top Curr Chem 225, 207-232 (2003)
- Clemmer et al., Annu Rev Anal Chem 1, 293-327 (2008)
- Gabelica and Marklund, Curr Opinion Chem Biol 42, 51-59 (2018)

FELIX





- 1) Background into IM-MS (what, history, fundamental principles)
- 2) Obtaining mobilities and CCS values
- 3) Instrumentation
- 4) Various types of ion mobility
- 5) Separation or characterization?
- 6) Applications / papers

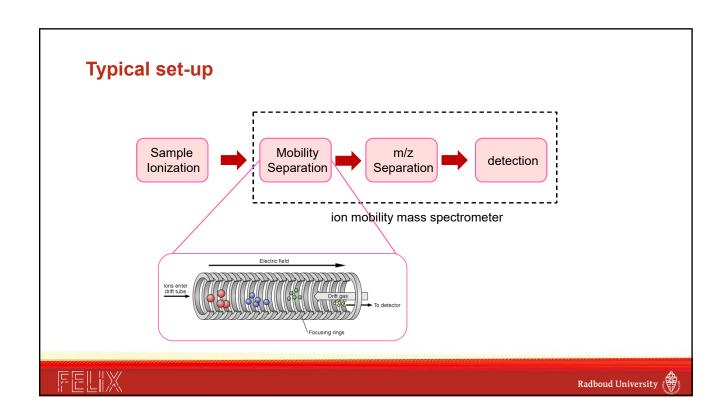
Background literature:

- Book: Ion Mobility Spectrometry- Mass Spectrometry – edited by Wilkens and Trimpin (2016)
- Wyttenbach and Bowers,
 Top Curr Chem 225, 207-232 (2003)
- Clemmer et al., Annu Rev Anal Chem 1, 293-327 (2008)
- Gabelica and Marklund, Curr Opinion Chem Biol 42, 51-59 (2018)

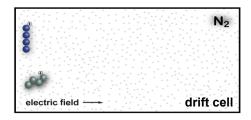
Special thanks to prof. Perdita Barran (University of Manchester) for sharing her IM slides For research by the Barran group: https://www.mbc.manchester.ac.uk/barrangroup/

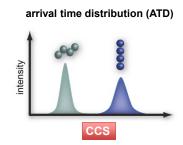






Ion Mobility: the general principle



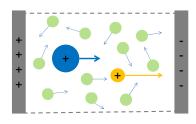


- · Gas-phase Separation of isobaric species
- Drift time converts to collision cross-section (CCS)
 Size information

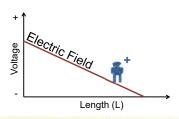
Radboud University



What is the basis for Ion Mobility separation?



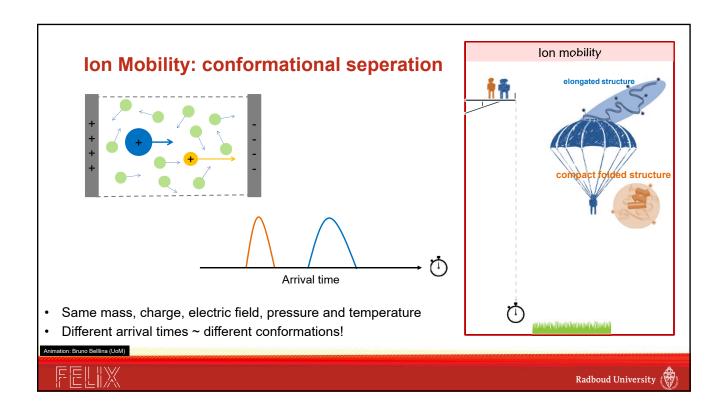
- · Electric Field "pulls"
- Gas Collisions "impede"
- Ratio of ion velocity (v) to the applied field (E) is Mobility (K)
- Applied over a length (L)

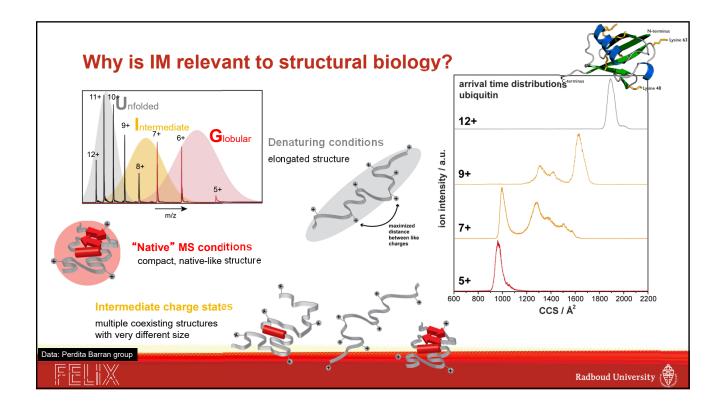


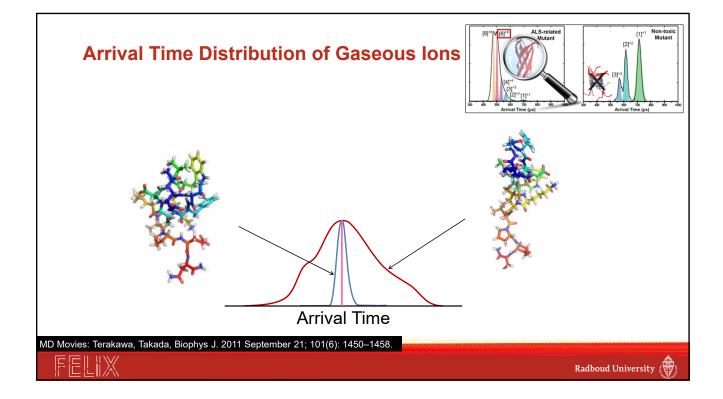
$$K = \frac{uL}{EE}$$











Historical Background

- · Which came first IMS or MS? Dates?
- Answer. IMS is ~ 15 years older than MS
- John Zeleny B.Sc. (1898) VI. On the ratio of the velocities of the two ions produced in gases by Röntgen radiation; and on some related phenomena - Philosophical Magazine Series 5, 46:278, 120-154, DOI:10.1080/14786449808621173

We are thus led to suppose, as in liquids, that the observed velocity difference is due to an inequality in the size of the two ions. Why the two ions, even if they are formed of groups of molecules, should in a simple gas be of a different size is a question to which definite answers cannot be given in the present state of our knowledge, or rather ignorance, of the relation between matter and electricity, but is one which must be borne in mind in considerations of this relation.

In conclusion, I desire to express my best thanks to Prof. J. J. Thomson for many valuable suggestions.

Cavendish Laboratory, April 12, 1898.







Timeline of developments in IM-(MS)

• 1890s J. Zeleny, J. J. Thomson, E. Rutherford

Interest in charge carriers gas after recent X-Ray discovery

1903 P. Langevin

Theoretical work; basis of todays theory

1913 F.W. Aston, J. J. Thomson

Development of first mass spectrometers

1950/60s Advances through military/security, pollution, and space research

Combination of Ion mobility and mass spectrometry (IM-MS) - study of ion-molecule reactions

Mason, McDaniel 1960/70s

Study of collision phenomena in ionized gases, book: "Transport Properties of Ions in Gases"

1990s Bowers, Kebarle, Hill, Russell

Metal ions and clusters. Later: biopolymers and biomolecules

Advances in structural biology (Jarrold, Clemmer, ...)

2007 1st commercial IM-MS setup (Waters Synapt HDMS)

since then larger molecules, complexes, many new instruments more accurate calculations, ...



Radboud University



Timeline of developments in IMS

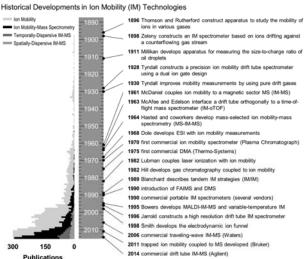


Figure 1. (Left) Histogram of the number of publications published per year in ion mobility and ion mobility-mass spectrometry. Note that the scale is truncated at 300 to highlight the number of publications specifically utilizing IM-MS. Further distinction is made to discriminate the frequency of publication for both time and space-dispersive IM-MS publications. (Right) Historical milestones in the development of ion mobility and IM-MS instrumentation.

May, McLean Analytical Chemistry 87, 1422 (2015)





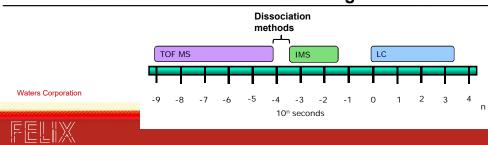
Applications of IMS



- Fundamental research: molecular structure, reactions, dynamic, kinetic
- Industry (pharmaceutical, semiconductor, petrochemical, ...): quality control / security, food security
- Medical applications: breath analysis, cancer recognition
- Security / military: Detection of narcotics, explosives, warfare agents
- Environmental monitoring



• IMS timescale: Combination with orthogonal methods



Radboud University



- 1) Background into IM-MS (what, history, fundamental principles)
- 2) Obtaining mobilities and CCS values -> Theory
- 3) Instrumentation
- 4) Various types of ion mobility
- 5) Separation or characterization?
- 6) Applications / papers

Background literature:

- Book: Ion Mobility Spectrometry- Mass Spectrometry – edited by Wilkens and Trimpin (2016)
- Wyttenbach and Bowers,
 Top Curr Chem 225, 207-232 (2003)
- Clemmer et al.,
 Annu Rev Anal Chem 1, 293-327 (2008)
- Gabelica and Marklund, Curr Opinion Chem Biol **42**, 51-59 (2018)

Special thanks to prof. Perdita Barran (University of Manchester) for sharing her IM slides For research by the Barran group: https://www.mbc.manchester.ac.uk/barrangroup/





Theory of IMS

• Definition of mobility K

 $v_{\rm d}$ – drift velocity

E-electric field

• Drift time t_d

L - drift length

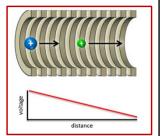
 t_0 – time offset

Ion mobility K

$$t_{\rm d} = \frac{L}{KE} + t_0$$

$$K = \frac{3}{16} \sqrt{\frac{2\pi}{\mu k_B T}} \frac{ze}{N\Omega}$$
 ccs

 $v_{
m d} = KE$ speed = $\frac{
m distance}{
m time}$



q – charge ze

N -drift-gas number density

 μ – reduced mass (ion + drift-gas molecule)

 $k_{\rm B}$ — Boltzmann constant

T - drift gas temperature

P-drift gas pressure







Theory of IMS

Definition of mobility K

 $v_{\rm d}$ – drift velocity

E – electric field

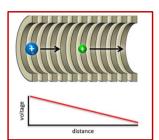
• Drift time t_d

L - drift length

 t_0 – time offset

Collision cross section Ω

 $t_{\rm d} = \frac{L}{KE} + t_0$



(Mason-Schamp equation)

 $\Omega = \frac{3q}{16N} \sqrt{\frac{2\pi}{\mu \ k_{\rm B}T}} \frac{1}{K_0}$

 $v_{
m d} = KE$ speed = $rac{
m distance}{
m time}$

q – charge ze

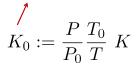
N - drift-gas number density

 μ – reduced mass (ion + drift-gas molecule)

 $k_{\rm B}$ — Boltzmann constant

T - drift gas temperature

P-drift gas pressure



reduced mobility





Theory of IMS – Ω determination

• Definition of mobility K

$$v_{\rm d} = KE$$

$$v_{
m d} = KE$$
 speed = $\frac{
m distance}{
m time}$

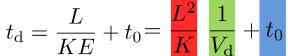
 $v_{\rm d}$ – drift velocity

E – electric field

• Drift time t_d

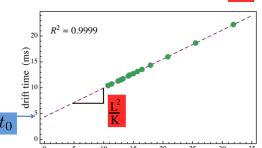
L - drift length

 t_0 – time offset









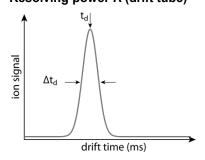


Arrival Time



Theory of Ion Mobility: Resolution

• Resolving power R (drift tube)



$$R_{
m DT} = rac{t_{
m d}}{\Delta t} = rac{1}{4} \sqrt{rac{q}{k_{
m B} \ln 2}} \sqrt{rac{V_{
m d}}{T}}$$

Example:

$$R_{\rm DT} = 100$$
$$\Delta\Omega = 1\%$$



- Increase R by increasing V_d or lower T
 - Limited by (low field condition)

 $\frac{1}{2}Mv_{\rm d}^2 \ll \frac{3}{2}k_{\rm B}T$



 $t_{\rm d} = \frac{L^2}{K} \frac{1}{V_{\rm d}} + t_0$

and electrical discharges





- 1) Background into IM-MS (what, history, fundamental principles)
- 2) Obtaining mobilities and CCS values
- 3) Instrumentation & various types of ion mobility
- 4) Separation or characterization?
- 5) Applications / papers



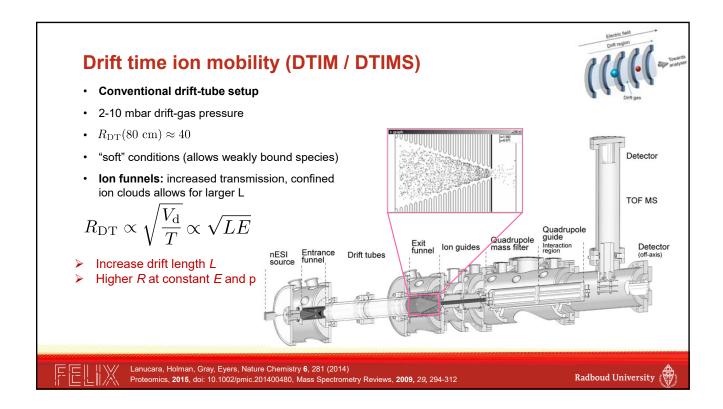
Radboud University

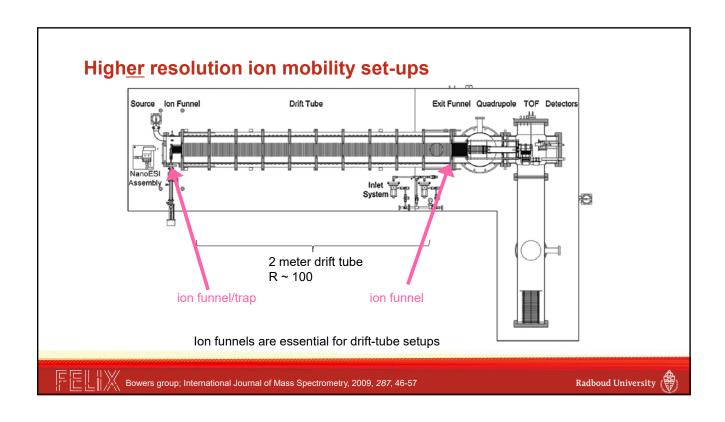


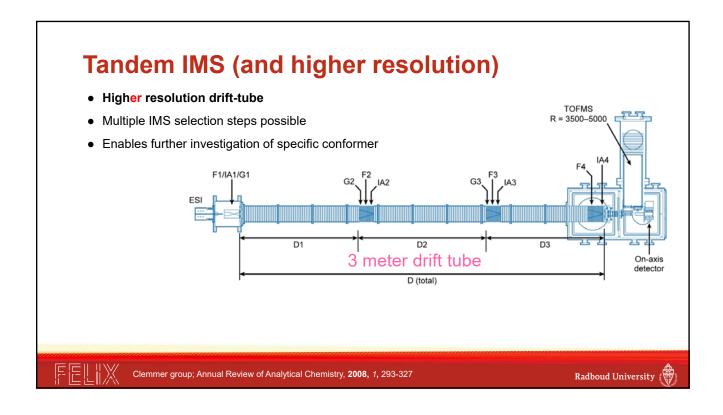
Drift Tube Ion Mobility

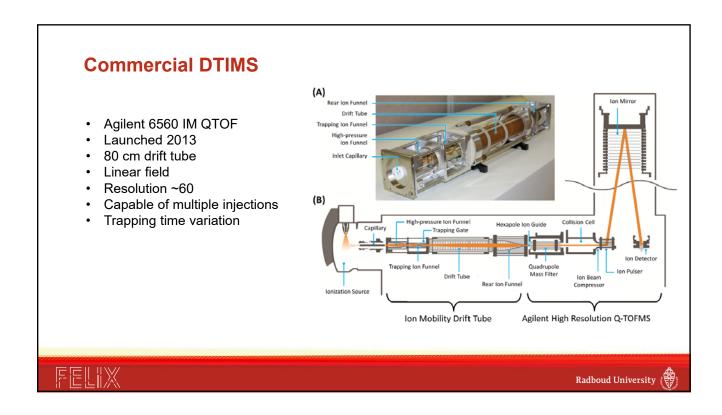












Traveling Wave Ion Mobility

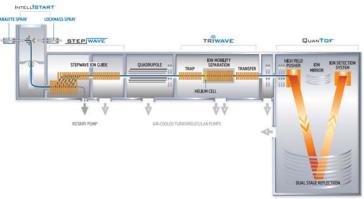






Commercial TWIMS: Synapt

- Developed by Waters
- · Ions radially confined by RF
- Buffer gas in drift cell within 'Triwave'
- $\Omega \propto t_D^x$
- Ions of high K roll over the wave less often tha species of low K = Shorter transit
- · Mobility 'based' separation
 - Commercially available
 - · High sensitivity
 - · High mass resolution
 - Pre- and post- IM activation possible
 - Ω determination requires calibration



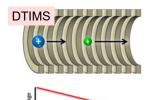
- Collision Cross Sections in Å² taken in N₂
- · Scaled to Helium CCS
- · Need for calibration

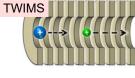






Travelling wave ion mobility (TWIMS)









- Series of ring electrodes (SRIG)
- · Electric field small region DT, wave pushes ion to detector
- Traveling wave upon which ion surf and traverse the IM cell
- Separation:
 - Higher mobility ions: carried by the wave
 - Lower mobility ions: roll over (longer time to move through)

SRIG

RF (+)

RF (-)

T-wave direction RF (+)

T-wave direction RF (+)

T-wave direction RF (+)

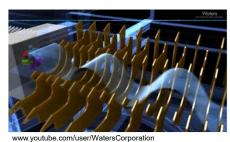
T-wave direction RF (+)

Biochimica et Biophysica Acta, 2011, 1811, 935-945 and Eyers et al., Nature Chemistry 6, 281 (2014)

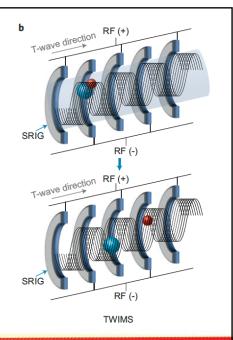


*Stacked Ring Ion Guide Radboud University

Travelling wave ion mobility (TWIMS)



- www.youtube.com/user/watersCorp
- Series of ring electrodes (SRIG)
- · Electric field small region DT, wave pushes ion to detector
- · Traveling wave upon which ion surf and traverse the IM cell
- · Separation:
 - · Higher mobility ions: carried by the wave
 - Lower mobility ions: roll over (longer time to move through)
- Use for separation and CCS characterization after calibration



Biochimica et Biophysica Acta, 2011, 1811, 935-945 and Eyers et al., Nature Chemistry 6, 281 (2014)



*Stacked Ring Ion Guide Radboud University

Travelling wave ion mobility (TWIMS)

TWIMS features:

· Compact design - high resolution

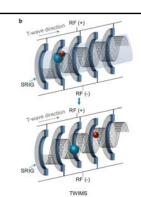
 $R_{\rm DT}(80~{\rm cm}) \approx 40$

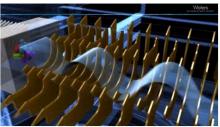
 $R_{\rm TW}(25~{\rm cm}) \approx 40$

No direct CCS (Ω) information

- non-uniform electric fields, not able to calculate CCS (Ω)
- Calibration required for Ω
- · Calibration requires defined conditions:
 - gas type and pressure
 - travelling wave speed and height
 - temperature
 - ion similar properties (size, charge, shape)

Biochimica et Biophysica Acta, 2011, 1811, 935-945 and Eyers et al., Nature Chemistry 6, 281 (2014)





www.youtube.com/user/WatersCorporation

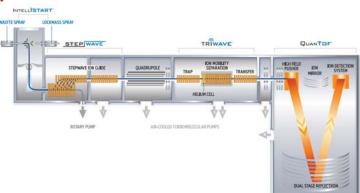






Commercial TWIMS: Synapt

- Developed by Waters
- · Ions radially confined by RF
- Buffer gas in drift cell within 'Triwave'
- Ω ∝ t_D^x
- lons of high K roll over the wave less often tha species of low K = Shorter transit
- · Mobility 'based' separation
 - Commercially available
 - High sensitivity
 - · High mass resolution
 - Pre- and post- IM activation possible
 - Ω determination requires calibration

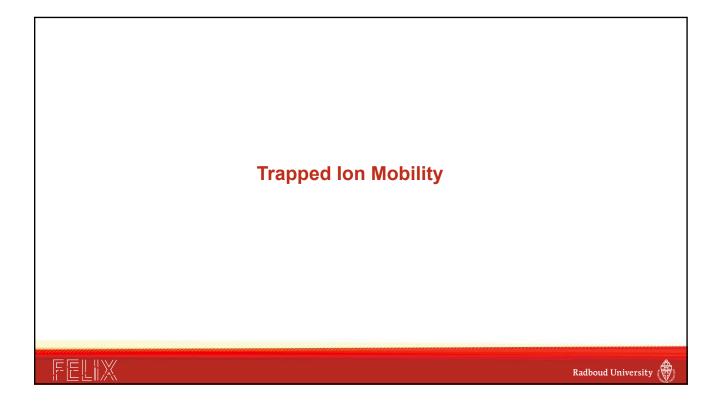


- Collision Cross Sections in Å² taken in N₂
- · Scaled to Helium CCS
- · Need for calibration

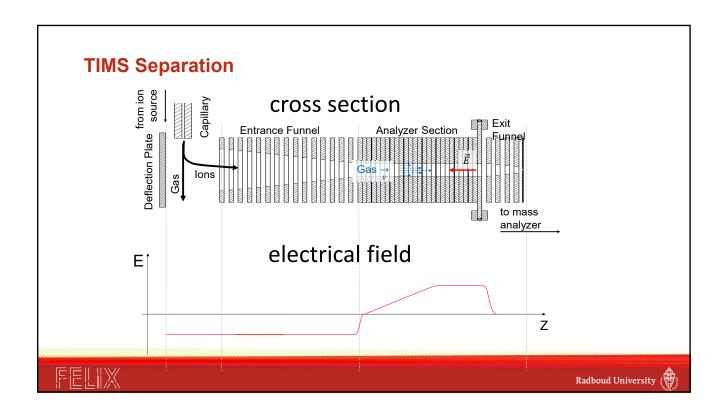


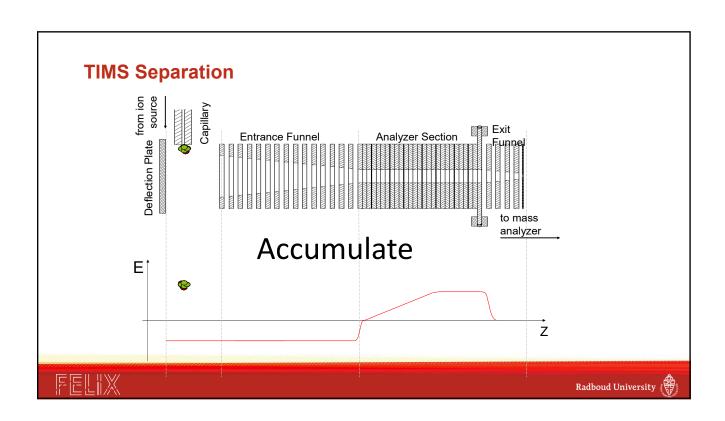


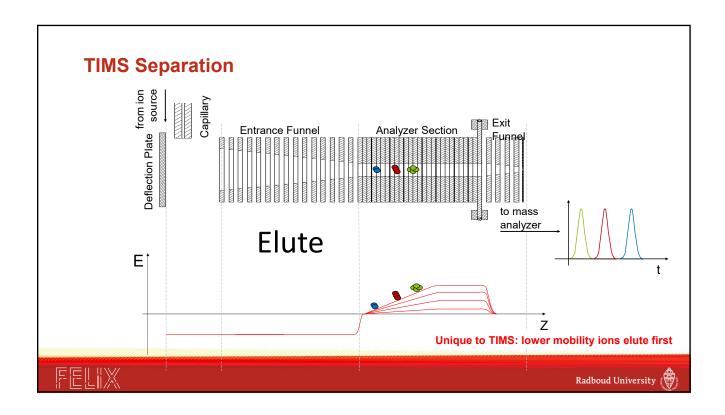


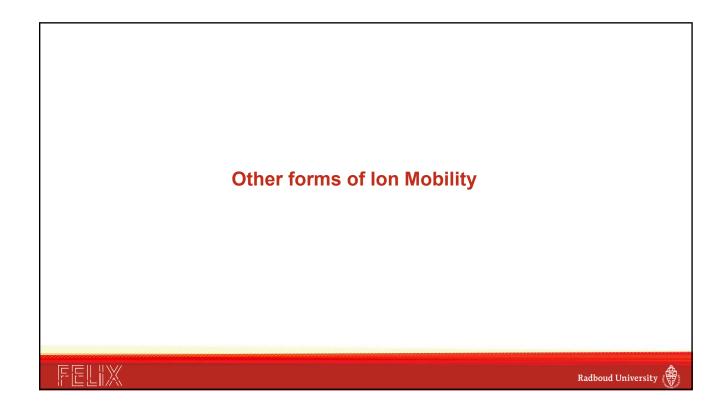


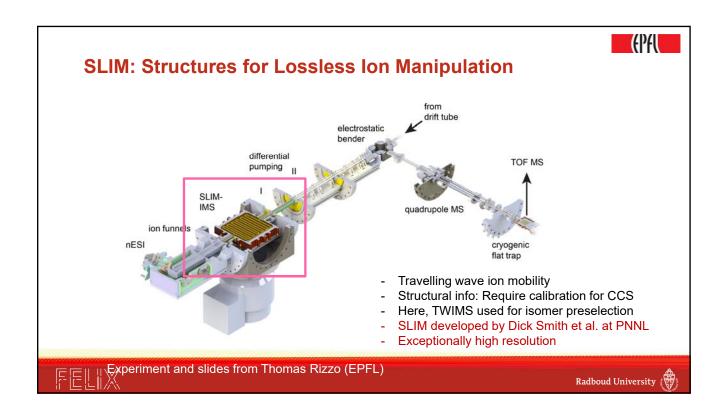


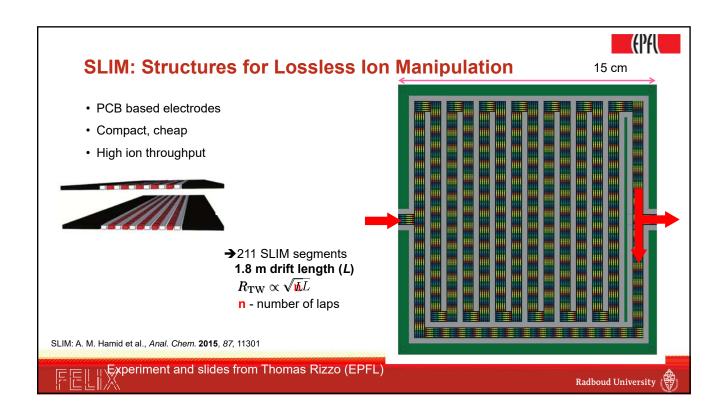






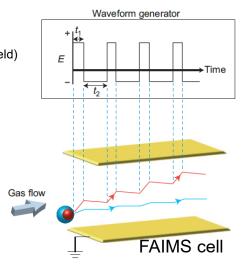






Field-asymmetric ion mobility (FAIMS)

- Ion introduced in alternating asymmetric field (low and high field)
- lons have different drift time towards the two electrodes
 - = different K of ions in low- and high electric fields
 - = separation
- FAIMS
 - High sensitivity
 - Separation of otherwise inseparable species
 - Possible to upgrade existing setup
- No CCS (Ω) information
- · Prediction very difficult
- "Hot" experimental conditions



FAIMS





REVIEW ARTICLE

NATURE CHEMISTRY DOI: 10.1038/NCHEM.1889

	DTIMS	TWIMS	FAIMS
Advantages	Rotationally averaged collisional cross- section (CCS; Ω), that is, 'shape' can be measured (Ų)	Rotationally averaged CCS can be determined	High resolving power (\le ~100 as defined by $\Omega/\Delta\Omega$ at FWHM) 14
		Can be used for mobility separation of	Relatively straightforward to transfer the ion mobility device between different mass spectrometers
	Can be used to separate species of very similar mobility; high resolving power (>100 as defined by $\Omega/\Delta\Omega$ measured at FWHM) ¹⁵⁷	product ions generated either by collision- induced dissociation or by electron-transfer dissociation	
Disadvantages	The geometric configuration of current commercial DTIMS-MS instruments means that they can only be used to separate analytes immediately post-ionization	CCS determination requires calibration of the drift time through the TWIMS cell, ideally using a calibrant of similar physical and chemical properties	CCS cannot be determined
			The geometric configuration of a FAIMS-MS instrument means that it can only be used to separate analytes immediately post-ionization
	Gating-type instruments are susceptible to ion losses when transferred from atmospheric pressure during ionization to the reduced pressure required for analysis	Relatively low resolving power (\leq -45 as defined by $\Omega/\Delta\Omega$ at FWHM) 152	
			The percentage of ions detected relative to those generated following ionization (that is, the duty cycle) is relatively low when operated under conditions where the CV is ramped (CV scanning mode), reducing sensitivity
		Ion heating can occur as ions are injected into the TWIMS cell which may affect gas-phase conformation. Unless carefully controlled, the process of measurement may therefore perturb analyte structure	

Eyers et al., Nature Chemistry **6**, 281 (2014)



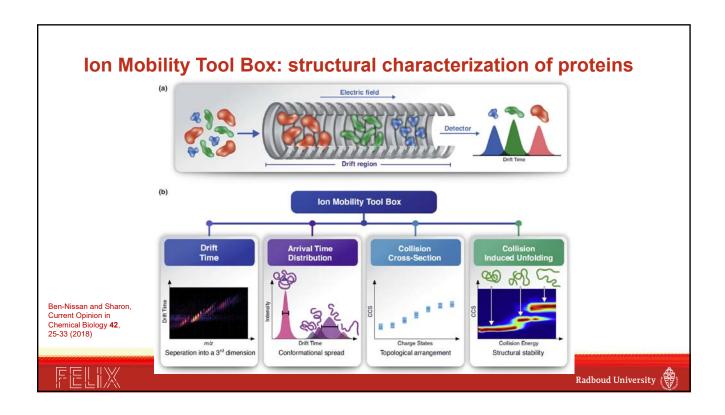


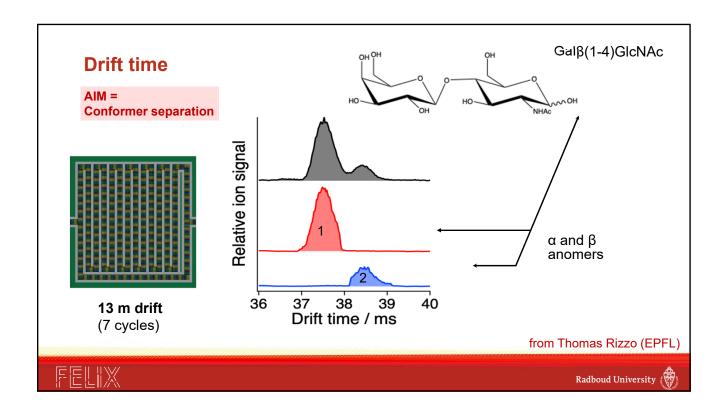
- 1) Background into IM-MS (what, history, fundamental principles)
- 2) Obtaining mobilities and CCS values
- 3) Instrumentation & various types of ion mobility
- 4) Separation or characterization?

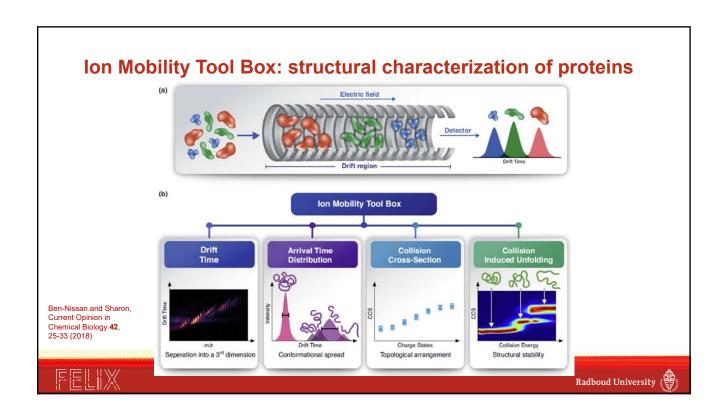
 Discussion in tutorial!
- 5) Applications / papers

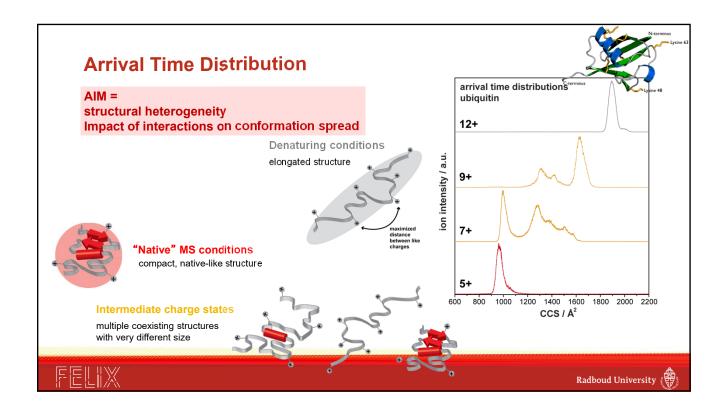


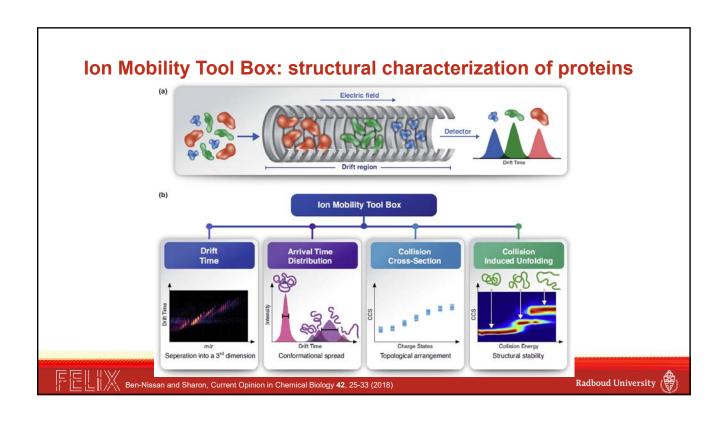


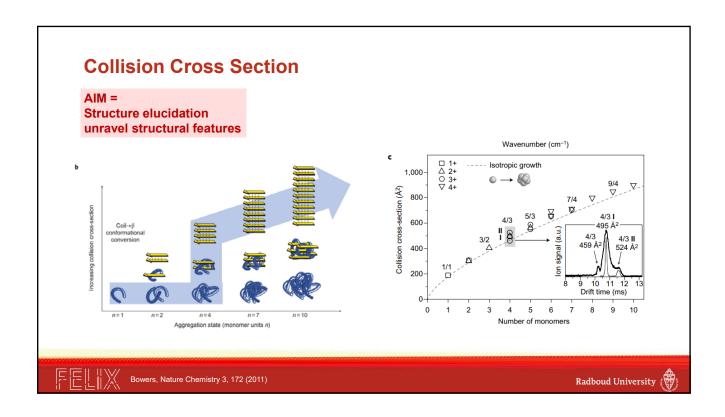


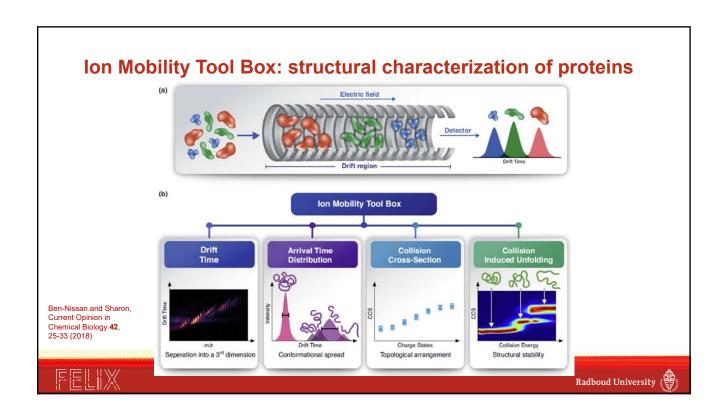












29+ 100 **Collision Induced Unfolding** — 80 V ■ % 50 - 40 V ■ - 8 V conformational stability Protein unfolding, organization 0 42 8V minimum voltage for ion transmission · native MS 40 Oriff time (ms) 36 40V MS: removal of salts and water (slightly lighter than blue) IM: protein collapses 80V 32 · MS: more removal water, salts IM values larger = 30 Unfolding of the protein complex 100 6250 6750 7250 7250 Justin Benesch, J.Am.Soc. Mass Spectrom 20, 341-348 (2009)

Summary Arrival Time Distribution 1) Background into IM-MS (what, history, fundamental principles) 2) Obtaining mobilities and CCS values = for drift tube via calculations Seperation into a 3rd dimension Conformational spread = for TWIMS (SLIM): via calibration 3) Instrumentation, types of ion mobility = DTIMS, TWIMS, SLIM, FAIMS, etc 4) Different experiments = drift time = arrival time distribution = collisional cross section Structural stability Topological arrangement = collisional induced unfolding Radboud University